

## SESQUITERPENE LACTONES AND OTHER CONSTITUENTS OF *ARNICA* *ACAULIS*

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**Key Word Index**—*Arnica acaulis*; Heliantheae; Compositae; helenanolides; melampolides; *trans*-fused guaianolides; sesquiterpene lactones; flavonoids.

**Abstract**—Chemical investigation of the aerial parts of *Arnica acaulis* resulted in isolation of 12 known helenanolides, one new melampolide baileyin acetate and the *trans*-fused guaianolides florilenalin acetate and 11 $\alpha$ H,13-dihydroflorilenalin acetate as well as other relatively common plant constituents. The implications for the biogenesis of helenanolides and the chemotaxonomic situation in *Arnica* are discussed.

### INTRODUCTION

*Arnica* (Compositae) is a circumboreal predominantly montane genus of about 32 species most of which are confined to western North America [1]. Its tribal position, traditionally within the Senecioneae, has been the subject of considerable discussion [2–4]. On morphological, serological and chemical grounds it has been argued that its affinities are with the Heliantheae rather than with the Senecioneae; thus the most recent treatment of Heliantheae [5] includes *Arnica* and its close relatives in subtribe Chaenactidinae which in turn appears to be closely related to subtribe Gaillardiiinae.

The chemical evidence which has been adduced in support of this relationship [5] includes *inter alia* the discovery [6–8] in *A. foliosa* Nutt. [*A. chamissonis* Less. ssp. *foliosa* (Nutt.) Maguire] and *A. montana* L. of sesquiterpene lactones of the helenanolide type which are characteristic constituents of Gaillardiiinae. Because of the role which *Arnica montana*, the only member of the genus found in Europe, plays in European popular medicine its chemistry and that of a few other *Arnica* taxa has been studied more thoroughly in recent years [9–22]. This has resulted in the isolation of an additional number of helenanolides with the biological effects of the drug being attributed primarily to helenalin (**1a**) and some of its derivatives.

The only *Arnica* species found in North America east of the Mississippi and south of the Adirondacks is *A. acaulis* (Walt.) B.S.P. According to Maguire [1] it has no near relative in North America while its general habitat and character of pubescence strongly suggest it to be an offshoot of European *A. montana*. It was therefore of interest to compare the chemistry of the two species. In the present report we describe isolation from *A. acaulis* of the sesquiterpene lactones **1a–e**, **2a–e**, **3**, **4**, **5b**, **6b** and **7b**, the

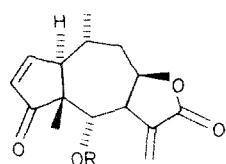
thymol derivative **9a** and the flavones pilloin (**10a**) and salvigenin (**11**).

### RESULTS AND DISCUSSION

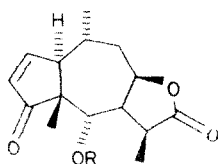
Extraction of the aerial parts of *A. acaulis* furnished a complex mixture of the known helenanolides **1a–e**, **2a–e**, **3** and **4** which were identified by comparison with authentic samples or by comparison with <sup>1</sup>H NMR spectra of esters of helenalin (**1a**) and 11 $\alpha$ H,13-dihydrohelenalin (**2a**) recorded in the literature. Lactones **3** (mexicanin I) and **4** (11 $\alpha$ H,13-dihydromexicanin I) have not been reported previously from *Arnica*.

A new crystalline lactone was the acetate **5b** of a C-8 lactonized melampolide baileyin (**5a**) from *Baileya plini-radiata* and *B. multiradiata* [23, 24], as shown by the <sup>1</sup>H and <sup>13</sup>C NMR spectra (Tables 1\* and 2), extensive decoupling and the observation of a significant NOE between H-1 and H-14 (Table 3). The H-2 $\beta$ , H-8 $\beta$ -stereochemistry originally suggested [24] for baileyin was speculative due to uncertainty about the conformations adopted by C-8 *cis*- and C-8 *trans*-lactonized melampolides. Since then it has been shown in the case of schkuhriolide and its derivatives [25–27] that the conformation of such C-8 *cis*-lactonized melampolides is similar to C-6 *trans*-lactonized members of this class, i.e. <sup>1</sup>D<sub>14</sub>, <sup>15</sup>D<sub>5</sub> [28], and gives rise to values of  $J_{7,13} < 3$  Hz, whereas C-8 *trans*-lactonized melampolides exemplified by frutescin [29, 30] assume the <sup>1</sup>D<sub>14</sub>, <sup>15</sup>D<sub>5</sub> conformation. Associated with this conformation of C-8 *trans*-lactonized melampolides are values of  $J_{7,13} > 3$  Hz, an upfield shift of H-8 to  $\delta$ 3.8–4.0 $\tau$  and relatively large values of  $J_{7,8}$  [29–31]. All this is clearly the case for **5a, b** which are therefore C-8 *trans*-lactonized. Inspection of the model constructed in accordance with Fig. 1 of ref [30] then requires  $\beta$ -orientation of the 2-hydroxy or acetoxy group to accommodate the values of  $J_{1,2}$ ,  $J_{2,3a}$  and  $J_{2,3b}$ ; moreover the model and the H-2 $\alpha$ , H-8 $\beta$  stereochemistry deduced from it for **5a, b** are in highly satisfactory agreement with the NOE difference spectrum of **5b**

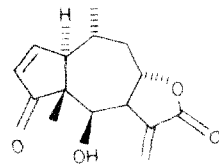
\*Table 1, column 1 of ref. [24] which lists the <sup>1</sup>H NMR spectrum of baileyin contains an erroneous assignment and several misprints. For this reason the correct NMR spectrum of **5a** is reproduced in Table 1.



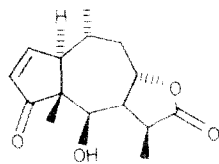
- 1 (a)** R  
**(b)** H  
**(c)** MeAcr  
**(d)** Tig  
**(e)** Sen  
**(f)** 2-Me Bu  
**(g)** Ac



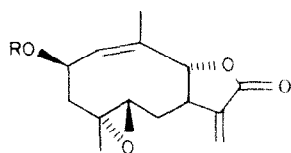
- 2 (a)** R  
**(b)** H  
**(c)** MeAcr  
**(d)** Tig  
**(e)** Sen  
**(f)** 2-Me Bu  
**(g)** Ac



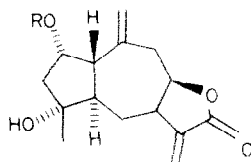
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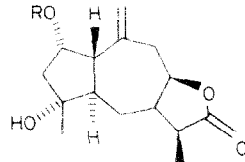
4



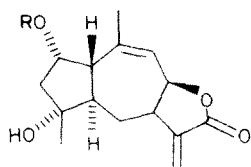
- 5 (a)** R  
**(b)** H  
**(c)** Ac



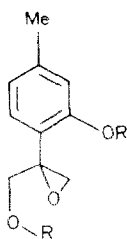
- 6 (a)** R  
**(b)** H  
**(c)** Ac



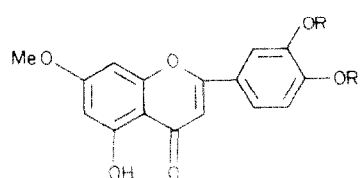
- 7 (a)** R  
**(b)** H  
**(c)** Ac



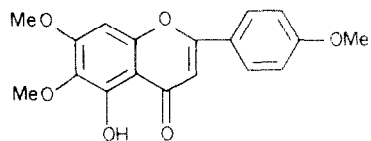
- 8 (a)** R  
**(b)** H  
**(c)** Ac



- 9 (a)** R  
**(b)** / Bu  
**(c)** / Val



- 10 (a)** R R'  
**(b)** H Me  
**(c)** Me H



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†A multiplet appearing at  $\delta 3.75$  in the  $^1\text{H}$  NMR spectrum of the C-8 *trans*-lactonized melampolide soulanganolide B has been ascribed to H-7[32]. The chemical shift of the H-8 signal was not reported. In the light of the preceding discussion it appears to us that the  $\delta 3.75$  signal of soulanganolide B is associated with H-8 and that the H-7 signal is found at higher field as usual.

detailed in Table 3. The CD curves of **5a** and **5b** exhibit positive maxima for the  $n, \pi^*$  transition of the  $\alpha, \beta$ -unsaturated lactone chromophore, hence the formulas also represent the absolute configurations (2*R*, 4*R*, 5*R*, 7*R*, 8*S*) of these compounds.

Table 1.  $^1\text{H}$  NMR data for compounds **5a**, **5b**, **6a** and **7b** (270 MHz,  $\text{CDCl}_3$ )

Hydrogen	<b>5a</b> *	<b>5b</b>	<b>6b</b>	<b>7b</b> ‡
1	5.33 <i>br d</i> (10)	5.28 <i>br d</i> (10)	1.93 <i>br d</i> (15)	1.90 <i>br d</i> (15)
2	4.69 <i>m</i> (10, 10, 6)	5.60 <i>dt</i> (6, 10)	5.29 <i>br dd</i> (6, 2.5)	5.16 <i>br t</i> (5.5)
3 $\alpha$	2.62 <i>dd</i> (12, 6)	2.58 <i>dd</i> (11, 6)	{ 2.20 <i>dd</i> (16, 6)	
3 $\beta$	1.27 <i>dd</i> (12, 10)	1.33 <i>t</i> (11)		
5	2.37 <i>br d</i> (12)	2.73 <i>dd</i> (10, 1)	{ ~2.06	
6 $\alpha$	2.33 <i>br d</i> (15)	2.39 <i>br d</i> (15)		
6 $\beta$	1.4 <i>m</i>	1.46 <i>dt</i> (15, 10)	{ ~2.10	
7	2.85 <i>m</i>	2.95 <i>m</i> (10, 6.5, 3.3, 2.9)		
8	4.05 <i>dd</i> (12, 9)	4.06 <i>ddd</i> (10.5, 6.5, 1.5)	4.62 <i>br ddd</i> (11, 8, 4)¶	4.55 <i>ddd</i> (11, 8, 3.5)
9 $\alpha$	2.47 <i>t</i> (12)	2.46 <i>dd</i> (13, 10.5)	{ 2.72 <i>br dd</i> (13, 4)¶	
9 $\beta$	2.79 <i>br d</i> (12)	2.79 <i>br d</i> (13)§		
13a	6.37 <i>d</i> (3.5)	6.35 <i>d</i> (3.3)	6.30 <i>d</i> (2.9)	2.50 <i>m</i> (H-11)
13b	5.67 <i>d</i> (2.7)	5.67 <i>d</i> (2.9)	5.69 <i>d</i> (2)	1.19 <i>d</i> (7)†
14	1.85 <i>br</i> (1)†	1.95 <i>br d</i> (1)†	5.07 <i>br</i>	5.11 <i>br</i>
15†	1.18	1.20 <i>br</i>	4.89 <i>br</i>	5.01 <i>br</i>
Ac†	—	2.03 <i>s</i>	1.22 <i>s</i>	1.19 <i>s</i>
			2.06s	2.08 <i>s</i>

\* Taken from ref. [24] with corrections.

† Intensity three protons.

‡ Taken from mixture with **6b**.

§ Long range coupled to H-14.

¶  $J_{6a,7}$  2Hz,  $J_{6b,7}$  = 12Hz.¶ Also coupled to H-14a at  $\delta$ 5.07.Table 2.  $^{13}\text{C}$  NMR data for compounds **5b** and **6b** (67.89 MHz,  $\text{CDCl}_3$ )

Carbon	<b>5b</b>	<b>6a</b>
1	127.87 <i>d</i>	53.34 <i>d</i> †
2	68.67 <i>d</i> *	73.89 <i>d</i>
3	46.93 <i>t</i> *	49.19 <i>t</i>
4	59.93 <i>s</i>	78.50 <i>s</i>
5	66.10 <i>d</i> *	51.69†
6	43.23 <i>t</i>	29.64 <i>t</i> *
7	46.23 <i>d</i>	39.66 <i>d</i>
8	83.61 <i>d</i> *	80.33 <i>d</i> *
9	30.98 <i>t</i> *	29.87 <i>t</i> *
10	138.94 <i>s</i>	139.94 <i>s</i> ‡
11	133.03 <i>s</i>	139.27 <i>s</i> ‡
12	168.63 <i>s</i>	169.68 <i>s</i>
13	121.81 <i>t</i>	122.54 <i>t</i>
14	18.18 <i>q</i> *	115.58 <i>t</i>
15	17.14 <i>q</i> *	25.32 <i>q</i>
Ac	170.16 <i>s</i>	170.55 <i>s</i>
	20.99 <i>q</i>	21.28 <i>q</i>

\* Assignment confirmed by single frequency decoupling.

†,‡ Assignments may be interchangeable.

Two additional non-crystalline lactones, the second obtained only in admixture with the first, were **6b** and its 11,13-dihydro-derivative **7b**. That these substances were C-8 lactonized guaianolides with a tertiary hydroxyl group on C-4 or C-10 and an exocyclic methylene group on C-10 or C-4 was evident from the  $^1\text{H}$  NMR spectra (Table 1), the  $^{13}\text{C}$  NMR spectrum of **6b** (Table 2) and

Table 3. NOE difference spectrum of compound **5b**

Saturation	Observed NOE (%)
H-1	3 $\beta$ (8.3) 14 (8.3) 15 (7) Ac (8.3)
H-2 and H-13b	3 $\alpha$ (11.4) 5 (11.4) 6 $\alpha$ (11.3) 9 $\alpha$ (11.3)
H-8	6 $\beta$ (8.2) 9 $\beta$ (8) 14 (12.3) 15 (8.2)
H-14	1 (6.7) 8 (13)
H-15	8 (8.9)

extensive decoupling in the usual way, beginning with irradiation at the frequencies of H-13a,b, which established the location of the H-7 signal and eventually led to formulation of the entire sequence C-3, C-2, C-1, C-6, C-7, C-8, C-9. As H-8 under the lactone oxygen and one of the two C-9 protons were both coupled to one of the two non-conjugated vinylic protons, the exocyclic methylene group was attached to C-10. The *trans*-fusion of the bicyclo(5,3,0) system followed from the very large coupling constants involving H-1 and H-5, with the remaining features of the stereochemistry also being deduced from the coupling constants. Thus **6b** was a double bond isomer of pleniradin acetate (**8b**), a derivative



mother liquors (hexane–Me<sub>2</sub>CO 4:1) afforded 115 mg of a 5:1 mixture of **1a** and **3** and mixtures containing **1a**, **2a**, **3** and **4** in varying proportions.

Rechromatography of frs 31–33 (375 mg) (20 ml fractions, Si gel, 30 g, eluent 100 ml each of Me<sub>2</sub>CO–CH<sub>2</sub>Cl<sub>2</sub> 1:19, 1:9 and 3:17) gave in frs 10–18 mainly florilenalin acetate (**6b**, 150 mg) and in frs 19–21 a 1:2 mixture of **6b** and **7b**. Rechromatography of crude **6b** gave pure florilenalin acetate as a gum (100 mg), <sup>1</sup>H and <sup>13</sup>C NMR in Tables 1 and 2; PCIMS *m/z* (%) 307 (*M*<sup>+</sup> + 1, 100), 289 (*M*<sup>+</sup> + 1 – H<sub>2</sub>O, 15.8), 229 (*M*<sup>+</sup> + 1 – H<sub>2</sub>O – C<sub>2</sub>H<sub>4</sub>O<sub>2</sub>, 33.9). The mixture of **6b** and **7b** could not be separated satisfactorily; the structure of **7b** was evident from the <sup>1</sup>H NMR spectrum (Table 1) and the PCIMS which exhibited additional strong peaks at *m/z* 309 (*M*<sup>+</sup> + 1), 291 and 231.

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#### REFERENCES

- Maguire, B. (1943) *Brittonia* **4**, 386.
- Herout, V. (1971) in *Pharmacognosy and Phytochemistry* (Wagner, H. and Hörhammer, L., eds) pp. 93–110.
- Turner, B. L. and Powell, A. M. (1977) in *The Biology and Chemistry of the Compositae* (Heywood, V. H., Harborne, J. B. and Turner, B. L., eds) pp. 699–737. Academic Press, London.
- Nordenstam, B. (1977) in *The Biology and Chemistry of the Compositae* (Heywood, V. H., Harborne, J. B. and Turner, B. L., eds) pp. 799–830. Academic Press, London.
- Robinson, H. (1981) *Smithsonian Contrib. Botany*, **51**, 1.
- Evstratova, R. I., Bankovskii, A. I., Sheichenko, V. I. and Rybalko, K. S. (1971) *Khim. Priro. Soedin.* **7**, 270.
- Poplawski, J., Holub, M., Samek, Z. and Herout, V. (1971) *Coll. Czechoslov. Chem. Commun.* **36**, 2189.
- Holub, M., Samek, Z. and Toman, J. (1972) *Phytochemistry* **11**, 2627.
- List, P. H. and Friebe, B. (1974) *Arzneim. Forsch.* **24**, 148.
- Willuhn, G. and Herrmann, H. D. (1976) *Arch. Pharm.* **309**, 333.
- Herrmann, H. D., Willuhn, G. and Hausen, B. M. (1978) *Planta Med.* **34**, 299.
- Willuhn, G. and Herrmann, H. D. (1978) *Pharm. Ztg.* **123**, 1803.
- Willuhn, G. and Herrmann, H. D. (1979) *Planta Med.* **37**, 325.
- Willuhn, G., Pretzsch, G. and Wendisch, D. (1981) *Tetrahedron* **37**, 773.
- Wiebecke, M., Kresken, J., Mootz, D. and Willuhn, G. (1982) *Tetrahedron* **38**, 2709.
- Willuhn, G., Kresken, J. and Wendisch, D. (1983) *Planta Med.* **47**, 157.
- Willuhn, G., Roettger, P.-M. and Mathiesen, V. (1983) *Planta Med.* **47**, 226.
- Willuhn, G., Roettger, P.-M. and Wendisch, D. (1984) *Planta Med.* **35**.
- Merfort, I. (1984) *Planta Med.* **107**.
- Willuhn, G., Junior, I., Kresken, J., Pretzsch, G. and Wendisch, D. (1985) *Planta Med.* **398**.
- Willuhn, G., Junior, I. and Wendisch, D. (1986) *Planta Med.* **349**.
- Merfort, I., Marcinek, C. and Eggert, A. (1986) *Phytochemistry* **25**, 2901.
- Waddell, T. G. and Geissman, T. A. (1969) *Phytochemistry* **8**, 2371.
- Herz, W., Murari, R. and Blount, J. F. (1979) *J. Org. Chem.* **44**, 1873.
- Samek, Z., Holub, M., Bloszyk, E. and Drozd, B. (1979) *Z. Chem.* **19**, 449.
- Rychlewska, U. (1982) *J. Chem. Soc. Perkin II*, 1641.
- Romo, de Vivar, A., Pérez, A. L., León, C. and Delgado, G. (1982) *Phytochemistry* **21**, 2905.
- Samek, Z. and Harmatha, J. (1978) *Coll. Czech. Chem. Commun.* **43**, 2779.
- Herz, W., Bhat, S. V. and Sudarsanam, V. (1972) *Phytochemistry* **11**, 1829.
- Herz, W., Prasad, J. S. and Blount, J. F. (1982) *J. Org. Chem.* **47**, 2206.
- Gutiérrez, A. B., Oberti, J. C., Sosa, V. E. and Herz, W. (1987) *Phytochemistry*, (in press).
- El-Ferali, F. S. (1983) *Phytochemistry* **22**, 2239.
- Yoshitake, A. and Geissman, T. A. (1969) *Phytochemistry* **8**, 1753.
- Pettit, G. R., Herald, C. L., Judd, G. F., Bolliger, G. and Thayer, P. S. (1975) *J. Pharm. Sci.* **64**, 2023.
- Lee, K. H., Ibuka, T., Kozuka, M., McPhail, A. T. and Onan, K. D. (1974) *Tetrahedron Letters* **2287**.
- Kozuka, M., Lee, K. H., McPhail, A. T. and Onan, K. D. (1975) *Chem. Pharm. Bull. Jpn* **23**, 1895.
- Herz, W. (1977) *Israel J. Chem.* **16**, 32.
- Watson, W. H. and Kashyap, R. P. (1986) *J. Org. Chem.* **51**, 2521.
- Bohlmann, F., Jakupovic, J., Ahmed, M. and Schuster, A. (1983) *Phytochemistry* **22**, 1623.
- Herz, W. and Högenauer, G. (1962) *J. Org. Chem.* **27**, 905.
- Metwally, M. A. and Dawidar, A. M. (1985) *Phytochemistry* **24**, 1377.
- Bohlmann, F., Niedballa, V. and Schulz, J. (1969) *Chem. Ber.* **102**, 864.
- Bohlmann, F., Misra, L. N. and Jakupovic, J. (1985) *Chemistry* **24**, 1378.
- Núñez-Alarcon, J. (1971) *J. Org. Chem.* **36**, 3829.
- Wollenweber, E. and Wassum, M. (1972) *Tetrahedron Letters* **797**.
- LeQuesne, P. W., Pastore, M. P. and Raiffauf, R. E. (1976) *J. Nat. Prod.* **39**, 391.
- Bondoni, A. L., Medina, J. E., Rondina, R. V. D. and Coussio, J. D. (1978) *Planta Med.* **34**, 328.
- Das, K. C., Farmer, W. C. and Weinstein, B. (1970) *J. Org. Chem.* **35**, 3989.